

## Volatile composition of *Gyrophragmium dunalii*

Sylvie Rapior<sup>1</sup>

Laboratoire de Botanique, Phytochimie et Mycologie,  
UM1/CNRS-UPR 9056, Faculté de Pharmacie,  
Université Montpellier 1, 15 avenue Charles Flahault,  
34060 Montpellier cedex 2, France

Marie-Josèphe Mauruc

Institut de Botanique, Service des Herbiers, Université  
Montpellier 2, 163 rue Auguste Broussonnet, 34090  
Montpellier, France

Jacques Guinberteau

INRA, Station de Recherches sur les Champignons, 71  
avenue Edouard Bourleaux, B.P. 81, 33883 Villenave  
d'Ornon Cedex, France

Christian-Louis Masson

Société d'Horticulture et d'Histoire Naturelle de  
l'Hérault, Section Mycologie, 163 rue Auguste  
Broussonnet, 34090 Montpellier, France

Jean-Marie Bessière

Laboratoire de Chimie appliquée, Ecole Nationale  
Supérieure de Chimie, 8 rue de l'École Normale, 34296  
Montpellier cedex 5, France

**Abstract:** *Gyrophragmium dunalii* was investigated for volatile compounds by GC/MS. Thirty volatile components were identified. The major volatile constituents of mushrooms from the Mediterranean basin and Atlantic Ocean sand dunes were benzaldehyde (44.2 and 28.0%, respectively) and benzyl alcohol (11.0 and 38.6%, respectively). The volatile component proportions correspond to 410 and 960  $\mu\text{g g}^{-1}$  of fresh weight from Mediterranean and Atlantic specimens, respectively. Combined benzaldehyde and benzyl alcohol content may contribute to the complex almond odor with an anise note of *G. dunalii*.

**Key Words:** Agaricaceae, almond-like odor, anise-like odor, Basidiomycota, benzaldehyde, benzyl alcohol, Podaxaceae

Higher fungi can generate a broad spectrum of odors (Maga 1981, Mazza 1998). Typical pleasant and unpleasant aromas of mushrooms are well known including bitter almond (Chen and Wu 1984), anise-

like (Wood et al 1990), fungal (Fischer and Grosch 1987, Rapior et al 1997b, Tressl et al 1982), garlic (Rapior et al 1997a), fenugreek (Rapior et al 2000), and musty-earthly (Breheret et al 1999) odors.

*Gyrophragmium dunalii* is a worldwide species growing on sand dunes (Guinberteau 1997, Kreisel 1973, Pilát 1958). An epigeous ochraceous fungus with a large annulus and typical funnel-shaped volva at the base of a very long hypogeous stipe, *G. dunalii* has dingy dark-brown anastomosing lamellae when old (Honrubia et al 1982, Moser 1978, Zeller 1943). According to authors, *G. dunalii* develops either *Agaricus bisporus* odor (Kreisel 1973) or cyanic (Pacioni 1982) and bitter almond smell with an anise-like note (Guinberteau 1999).

As far as we know, this species was first considered as a hymenomycetous species and named *Agaricus ocreatus* (Raffeneau-Delile 1820–1840). According to Montagne (1843), Fries named the mushroom as *Montagnites delilei* but published the species as *M. dunalii* (Fries 1836–1838). Then Montagne (1843) classified the fungus as *Gyrophragmium delilei* (type genus) into the tribe Podaxinées (Gasteromycetes) using macroscopic characteristics of young basidiomes from Algeria. Later Zeller (1943) called it *G. dunalii* (Fr.) Zeller according to International Rules of Botanical Nomenclature. Recently, according to Hawksworth et al (1995) and Honrubia et al (1982), *G. dunalii* belongs either to Agaricales Clements (*Podaxaceae* Corda, secotioid species) or Podaxales Zeller (*Secotiaceae* Tul., agaricoid species), respectively.

The present work is the first chemical study on the complex almond odor with an anise note of *G. dunalii*. The fresh mushroom was investigated for volatile components by solvent extraction using gas chromatography/mass spectrometry (GC/MS). Thirty-one and 56 g of fresh *G. dunalii*, representing a combination of young and mature sporophores, were collected on sand dunes from the Mediterranean basin in Dec 1998 in the National Reserve of Camargue (France) and Atlantic Coast in Nov 1999 in the Oléron Island (France), respectively. The specimens were brushed clean, cubed and treated immediately after collection with diethyl ether (120 mL and 220 mL, respectively) to stop enzymatic activity. The organic extracts were gently concentrated to a small volume (0.5 mL) under nitrogen stream and analyzed (1.0  $\mu\text{L}$ ) in duplicate by GC/MS.

Solvent extract analyses from fresh *G. dunalii* were carried out with a gas chromatograph (5890-Hewlett-Packard) and a mass selective detector (5971-Hewlett-Packard) with

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<sup>1</sup> Email: srapior@ww3.pharma.univ-montpl.fr

TABLE I. Volatile composition of fresh *Gyrophragmium dunalii*

Volatile compounds	RI <sup>a</sup>	Location			
		Mediterranean		Atlantic	
		% <sup>b</sup>	$\mu\text{g g}^{-1\text{c}}$	% <sup>b</sup>	$\mu\text{g g}^{-1\text{c}}$
3-Methylbutanol	716	0.4	1.6	0.1	1.0
2-Methylbutanol	720	0.3	1.2	—	—
Hexanal	797	0.6	2.4	0.2	2.0
3-Methylbutanoic acid	846	0.6	2.4	0.2	2.0
2-Methylbutanoic acid	854	0.2	0.8	—	—
Styrene	870	0.4	1.6	—	—
1-Hexanol	904	0.1	0.4	—	—
Benzaldehyde	953	44.2	181.0	28.0	273
Methylstyrene <sup>d</sup>	977	0.2	0.8	0.1	1.0
2- <i>n</i> -Pentylfuran	985	0.3	1.2	—	—
Phenol	999	0.3	1.2	0.3	2.9
Benzyl alcohol	1036	11.0	45.0	38.6	376
2-Phenylethanal	1037	2.0	8.0	3.1	30.7
Acetophenone	1060	0.4	1.6	—	—
<i>n</i> -Octanol	1087	0.1	0.4	0.1	1.0
Nonanal	1098	0.5	2.0	—	—
Undecane	1100	0.1	0.4	—	—
2-Phenylethanol	1112	2.1	8.6	6.2	61.4
Benzoic acid	1197	10.5	43.0	7.0	69.3
Phenylethanoic acid	1254	0.6	2.4	0.3	3.0
2-Undecanone	1294	0.1	0.4	—	—
Tridecane	1300	0.6	2.4	—	—
Indole	1303	1.5	6.1	0.4	4.0
Benzamide	1342	1.8	7.4	—	—
<i>p</i> -Hydroxybenzaldehyde	1398	0.6	2.4	—	—
<i>p</i> -Aminobenzaldehyde <sup>e</sup>	1420	8.0	33.0	4.8	47.5
2-(2'-thienyl)-thiophene	1432	0.5	2.0	—	—
Cinnamic acid	1453	0.7	2.9	6.0	59.4
Unidentified	1481	0.5	2.0	0.3	3.0
Pentadecane	1500	0.7	2.9	—	—
Unidentified	1567	4.9	20.0	1.2	11.9
3-Carboxyindole	1758	4.0	17.0	1.2	11.9

<sup>a</sup> Retention indices on polydimethylsiloxane Optima 5 column.

<sup>b</sup> Percentage of total ion current (TIC).

<sup>c</sup> Amount of volatile compounds ( $\mu\text{g g}^{-1}$  of fresh weight).

<sup>d</sup> Unidentified isomer.

<sup>e</sup> Tentatively identified.

a potential of 70 eV for ionization by electron impact. GC/MS analyses were performed by a polydimethylsiloxane Optima 5 (Macherey-Nagel, 25 m length, 0.20  $\mu\text{m}$  i.d., 0.13  $\mu\text{m}$  phase thickness), fused silica capillary column (J & W). The carrier gas was helium with a linear gas velocity of 28  $\text{cm s}^{-1}$ . The injector and detector temperatures were 200 C and 270 C, respectively. The column was temperature programmed as follows: 50 C (3 min) to 200 C (3 C/min).

Thirty volatile components were identified by GC/MS (TABLE I). The volatile compound proportion

corresponds to 410 and 960  $\mu\text{g g}^{-1}$  of fresh weight for Mediterranean and Atlantic locations, respectively. Retention indices and experimental mass spectra of volatiles were compared to those available from literature spectra (Adams 1989, Jennings and Shibamoto 1980), mass spectral library NBS (MacLafferty and Stauffer 1989), and our own data bank. There is no significant qualitative difference from the major volatile compounds of specimens collected from the Mediterranean Sea and Atlantic Ocean dunes. But quantitative differences in volatile components are reported in TABLE I: the main difference is relative to benzyl alcohol content (11.0% and 45.0  $\mu\text{g g}^{-1}$ , and 38.6% and 376  $\mu\text{g g}^{-1}$ , respectively).

The main volatile compounds of *G. dunalii* were aromatic derivatives classified in two groups. Benzylic derivatives (benzaldehyde, benzyl alcohol, benzoic acid, benzamide, *p*-hydroxybenzaldehyde, *p*-aminobenzaldehyde) represented 76.1 and 84.9% of the volatile substances listed in TABLE I, while phenylethyl derivatives (2-phenylethanal, 2-phenylethanol, phenylethanoic acid) constituted only 4.7 and 10.2% of the volatile fraction from Mediterranean and Atlantic specimens, respectively.

The major volatile components of *G. dunalii* were quantitatively benzaldehyde (44.2 and 28.0%, respectively) and benzyl alcohol (11.0 and 38.6%, respectively) as well as the volatile compound (RI: 1420, 8.0 and 4.8%, respectively) tentatively identified as *p*-aminobenzaldehyde. The experimental mass spectrum (ms) of the latter matches the ms expected according to its chemical structure and is identical to the ms published for *m*-aminobenzaldehyde (MacLafferty and Stauffer 1989). The calculated retention indices of the three aminoacetophenone isomers confirm the para structure of the aminobenzaldehyde component. No odor description was reported for the latter volatile. It is not commercially available due to its instability.

Among the flavor volatile components, many aliphatic alcohols (3-methylbutanol, 2-methylbutanol, 1-hexanol, *n*-octanol) and aldehydes (hexanal, nonanal) were identified from *G. dunalii* (TABLE I). Other minor fragrant components, i.e., indole and phenol, with low odor threshold may significantly contribute to fungus aroma despite low contents. However, the similarity between aroma specimens from Mediterranean and Atlantic sites leads us to think that the main volatile constituents (benzaldehyde and benzyl alcohol) also characterized by low odor thresholds have a major contribution to the complex aroma of *G. dunalii*.

Investigations show that a close and demonstrable affinity exists between certain *Secotiaceae* and certain Agaricales (Singer 1958). An evaluation of *Gyrophrag-*

*mium* microscopic characters supports its inclusion in the *Agaricaceae* as a secotioid genus (Kreisel 1973). According to Guinberteau (1999), the positive Schaeffer reaction on the epicutis of *G. dunalii* as well as the almond/anise-like odor exhibit demonstrable affinities with agarics especially from the section *Arvenses* Konrad & Maublanc. Wood et al (1990) identified benzaldehyde (almond odor) and benzyl alcohol (sweet-spicy odor) as well as hydroxybenzaldehyde isomers (sweet-woody balsamic odor) in *Agaricus augustus*. The authors demonstrated the odor of benzaldehyde and benzyl alcohol mixtures as to be either almond-like or anise-like depending on the relative concentrations of these two chemicals. High contents of benzaldehyde and benzyl alcohol were also reported in *A. bisporus* (Fischer and Grosch 1987), *A. campestris* (Buchbauer et al 1993), and *A. subrufecens* (Chen and Wu 1984).

Benzaldehyde and benzyl alcohol may contribute to the specific almond odor with an anise note of fresh *G. dunalii*. Benzaldehyde with its almond odor is widely used as an aroma component in food, cosmetics and flavor industries. Considering consumer preference for natural aroma components, the potential for biotechnological production of volatiles from the Basidiomycota is currently being developed as reported in the literature (Lomascolo et al 1999, Schindler and Schmid 1982).

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